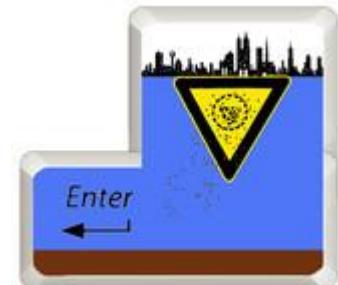


Interlaboratory comparison study on AuNPs

**WG 2 – Analytical Toolbox
COST Action ES1205 – ENTER**

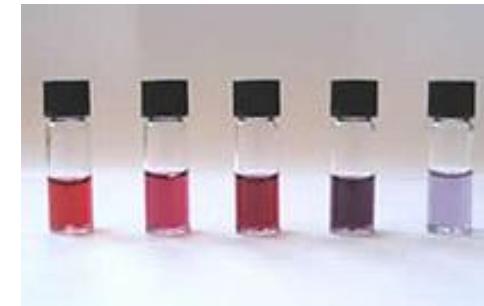
in cooperation with

NORMAN WG4 Engineered Nanomaterials



Tasks

- **5 AuNPs samples were distributed**
 - * three monodmodal samples
 - * two mixtures
 - * one blank



- tasks:
 - * determination of particle sizes / concentration
 - * determination of total Au concentration
 - * based on EU definition: nano – yes/no?!

Thanks to NORMAN for funding



Attendees

1 - Tommaso Serchi & Sébastien Cambier
CRP Gabriel Lippmann – Luxembourg

2 - Geert Cornelis
Gothenburg University – Sweden

3 - Irena Ciglenečki – Jušić & Marija Marguš
Institute Ruđer Bošković – Croatia

4 - Allan Philippe
Institute for environmental sciences, University of Koblenz/Landau – Germany

5 - Thomas Bucheli & Alexander Gogos
Institut Agroscope – Switzerland

6 - Frank v.d. Kammer & Stephan Wagner
University of Vienna – Austria

7 - Martin Tröster
DVGW-Technologiezentrum Wasser (TZW) – Germany

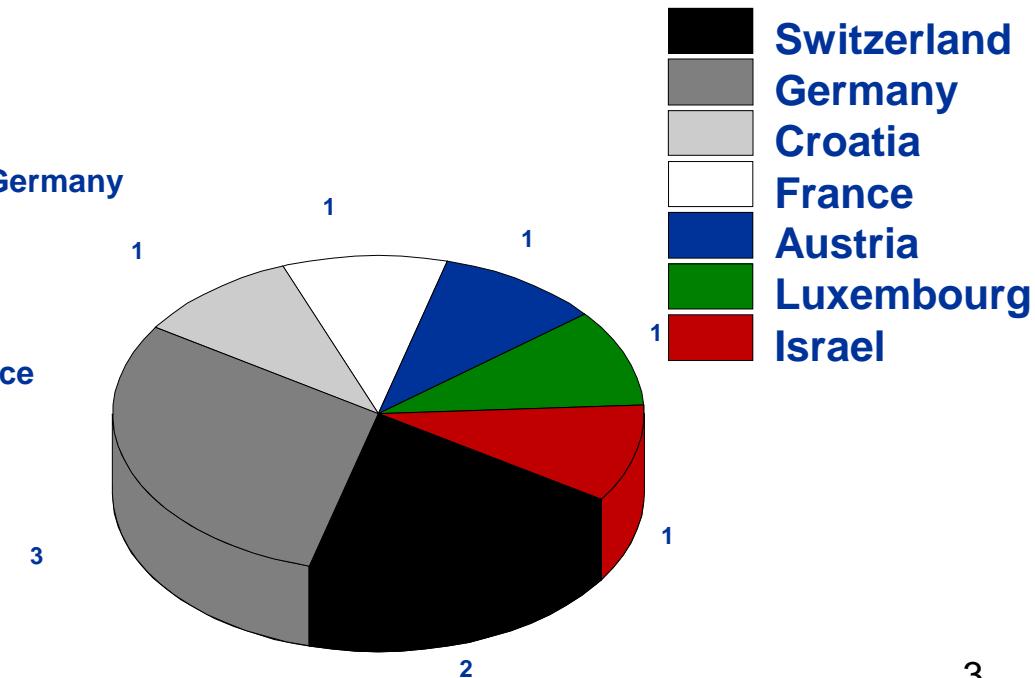
8 - Ralf Kaegi
Eawag – Switzerland

9 - Mickaël Tharaud & Yann Sivry
Institut de Physique du Globe de Paris – France

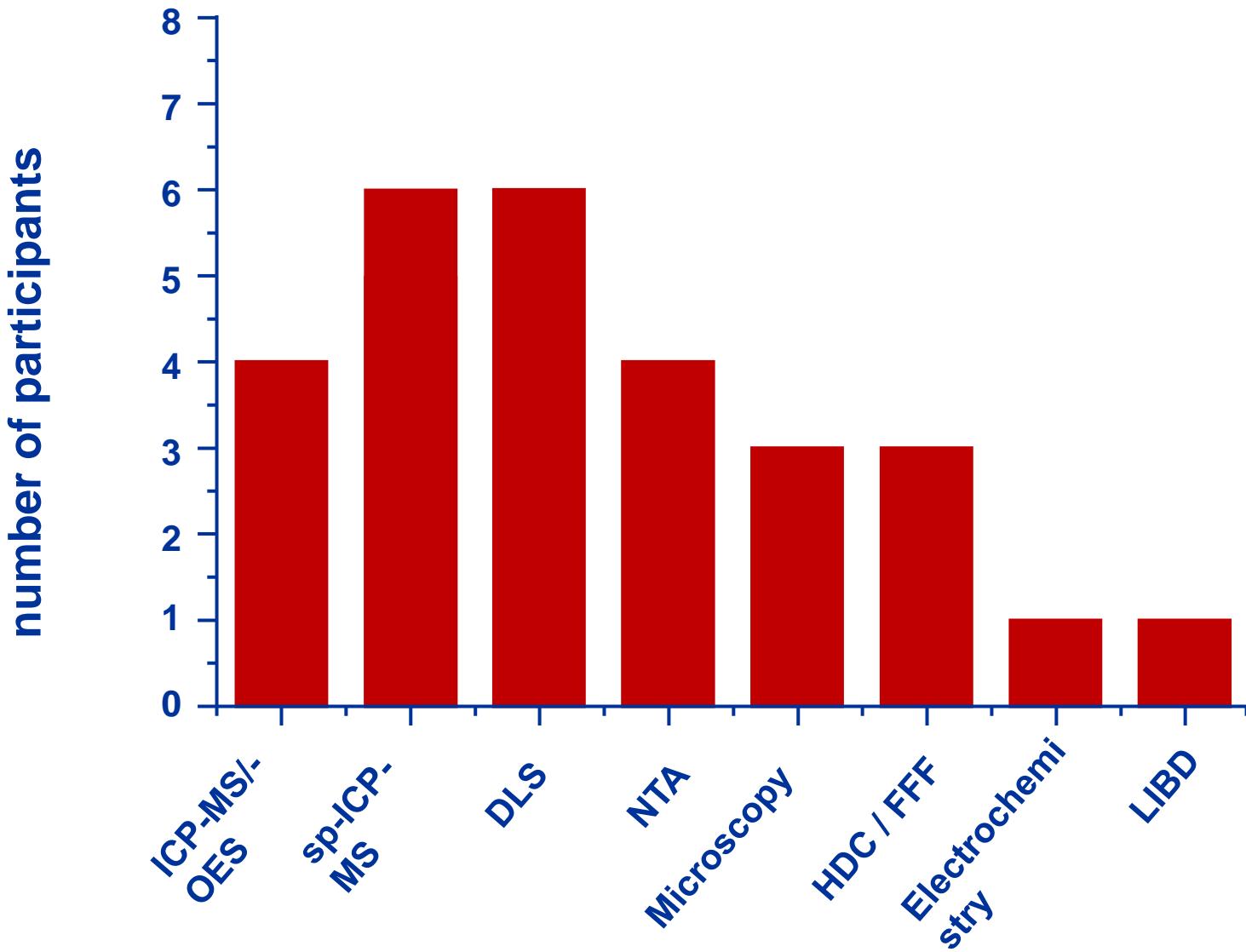
10 - Ilit Cohen-Ofri
Institute of Science – Israel

11 - Rute Isabel Ferreira Domingos
Instituto Superior Técnico – Portugal

12 - Björn Meermann
Federal Institute of Hydrology – Germany#



Overview – techniques deployed



Standards and Samples

- Standards

- * Au NP standards from BBI (UK)
- Thanks to NORMAN for funding
- * three different sizes
- * supplier information:



Material	D [nm]	V _{Std} [L]	d[nm] ¹⁾	c [#/mL] ¹⁾	c [mg Au/L] ¹⁾
Au in citrate	10	1.0	9.5	5.7×10 ¹²	57.6
Au in citrate	50	1.0	49.8	4.5×10 ¹⁰	56.8
Au in citrate	250	1.0	253.5	3.6×10 ⁸	56.8

1) No information on measurement method applied by supplier 5

Standards and Samples

- pre-treatment

* Shaking

- flasks and vials

* New, unused and cleaned 0.5 L and 1 L borosilicate glass flasks

* New, unused and cleaned 20 mL borosilicate glass vials

- monodisperse samples (samples 1 - 3)

* Standards “as obtained”

Sample Nr.	Nominal Size [nm]	Concentration [particles/mL]	Concentration [mg Au/L]
1	250	3.6×10^8	56.8
2	50	4.50×10^{10}	56.8
3	10	5.70×10^{12}	57.6

Standards and Samples

- Polymodal samples (samples 4 + 5)

* Mixtures of different size standards

Sample 4					
	Nom. Size [nm]	Vol. Fraction	Conc. in mix. [particles/mL]	Part-fract. in mix. [nr-%]	Conc. in mix. [mg Au/L]
Standard 10 nm	10	0.3	1.71×10^{12}	98.7	17.28
Standard 50 nm	50	0.5	2.25×10^{10}	1.3	28.4
Standard 250 nm	250	0.2	7.20×10^7	0.004	11.36
Total			1.73×10^{12}		57.04

→ Nanomaterial

Sample 5					
	Nom. Size [nm]	Vol. Fraction	Conc. in mix. [particles/mL]	Part-fract. in mix. [nr-%]	Conc. in mix. [mg Au/L]
Standard 10 nm	10	0.0	0	0	0
Standard 50 nm	50	0.05	2.25×10^9	86.8	2.84
Standard 250 nm	250	0.95	3.42×10^8	13.2	53.96
Total			2.59×10^9		56.8

→ Nanomaterial

- Blank (milliQ) (sample 6)

Results - NTA

	Ref		1		4		11		12	
No.	#/mL	d [nm]	#/mL	d [nm]	#/mL	d [nm]	#/mL	d [nm]	#/mL	d [nm]
1	3.6×10^8	250	$4.50E^8$	243	$2.81E^8$	242	Agg.	Nd	$1.26E^9$	274
2	4.50×10^{10}	50	$3.31E^{11}$	47	$6.16E^{10}$	54	$1.00E^{11}$	49 99	$2.41E^{10}$	63
3	5.70×10^{12}	10	$2.30E^8$	48	$3.22E^8$	114	$1.60E^{10}$	104 180	$1.20E^9$	96
4	1.71×10^{12} 2.25×10^{10} 7.20×10^7	10 50 250	$1.69E^{11}$	46 100 /200	$6.83E^{10}$	50	$5.60E^{10}$	51 100	$7.14E^8$	60
5	2.25×10^9 3.42×10^8	50 250	$1.76E^{10}$	48 200/300	$3.77E^9$	240	$3.70E^{10}$	102 186	$2.30E^8$	259

Monomodal: ok, but size DL > 10 (Au)
Polymodal: poor

Results – sp-ICP-MS I

	Ref		2		4		6		7	
No	#/mL	d [nm]	#/mL	d [nm]	#/mL	d [nm]	#/mL	d [nm]	#/mL	d [nm]
1	3.6×10^8	250	1.25E⁶	45	-	271	$1.40E^8$	322	1.40E⁸	271
2	4.50×10^{10}	50	1.13E¹²	47	-	49	$4.40E^{10}$	58	5.20E¹⁰	48
3	5.70×10^{12}	10	5.01E¹¹	11	-	46	$1.60E^{11}$	27	Nd	Nd
4	1.71×10^{12}	10	9.76E¹¹	13	-	40	$4.85E^9$ $1.74E^{10}$	32/60	$9.6E^{10}/$ $2.2E^{10}/$ $3.80E^7$	22/48/ 260
5	2.25×10^9	50	6.13E⁹	45	-	60	$4.10E^9$ $1.40E^9$ $4.9E^8$	27 60 315	$4.8E^8/$ $1.9E^9/$ $1.5E^8$	29/47/ 257
	3.42×10^8	250		60						

... depends on the instrument

Results – overview

Results – general overview

Method	NTA; N=4		sp-ICP-MS; N=6		DLS; N=6		AFM/TEM; N=3	
Sample	average Part/mL	RSD [%]	average Part/mL	RSD [%]	average Size [nm]	RSD [%]	average Size [nm]	RSD [%]
1	6.63E ⁸ 253	79 7	1.68E11 201	223 60	300	56	273	11
2	1.29E ¹¹ 55	107 14	7.63E13 49	223 9	45	19	48	8
3	4.44E ⁹ 86	174 39	4E15 25	173 63	12	25	9	13
4	7.34E ¹⁰ 55	95 12	Nd	Nd	52	32	Nd	Nd
5	1.46E ¹⁰ 250	114 5	Nd	Nd	247	23	Nd	Nd
6	Blank	Blank	Blank	Blank	Blank	Blank	Blank	Blank

Conclusions

- **NTA, DLS** good comparability for monodisperse samples (size);
polydisperse samples not all fractions detectable (known issue)
- **sp-ICP-MS** in general good comparability (size; number concentration
at least same order of magnitude)
but: strong dependency on instruments
- **AFM, TEM** comparable results for monodisperse samples (size)
but: polydisperse samples strongly influenced by sample
preparation; number concentration difficult to achieve
- **AF4, HDC** good agreement for monodisperse as well as polydisperse
samples
but: determination of number concentration laborious

'Micro-' meets 'Nano-' particles: 'Inclusion' of the Microplastics into the WG4

Microplastic and Nanoparticles are both particulate 'contaminants' and thus require tailored sample preparation protocols to preserve the particle size / number (mass is conservative, number not!)

The (quantitative) analytics is very challenging in complex matrices (sewage sludge, sediments) and not routinely established

First joint MP-NP-activities:

SETAC 2016: Session: Detection, identification and quantification of engineered (nano)materials in complex matrices

Patrick Bäuerlein (KWR), Ralf Kaegi (Eawag)

This session aims to integrate contributions focussing on latest developments for **nanoparticle and microplastics** analysis including explicitly sample preparation techniques...